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Structure of (3R,6R,7R,11S)-7-Acetoxymethyl-5,8,10-trioxa-1-azatricvclo[4.3.2.0^{3,11}]undecan-4-one

BY KINGA SUWIŃSKA

Institute of Physical Chemistry, Polish Academy of Sciences, Kasprzaka 44/52, 01 224 Warsaw, Poland

AND IRMA PANFIL, CZESŁAW BEŁŻECKI AND MAREK CHMIELEWSKI

Institute of Organic Chemistry, Polish Academy of Sciences, Kasprzaka 44/52, 01 224 Warsaw, Poland

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Abstract. $C_{10}H_{13}NO_6$, $M_r = 243.2$, monoclinic, $P2_1$, a $= 5.558 (2), \quad b = 8.119 (4), \quad c = 11.947 (4) \text{ Å}, \quad \beta = 11.947 (4) \text{ Å},$ $98.19 (2)^{\circ}, V = 534 (2) \text{ Å}^3, Z = 2, D_x = 1.514 \text{ g cm}^{-3},$ λ (Mo K α) = 0.71073 Å, μ = 1.185 cm⁻¹, F(000) = 256, T = 298 K, final R = 0.046 for 920 observed reflections. The two five-membered rings adopt different conformations: envelope and half-chair. The seven-membered ring shows the distorted chair conformation. The shortest non-bonded contact is 3.29 Å for non-H atoms.

Experimental. Crystal of approximate dimensions $0.28 \times 0.28 \times 0.30$ mm; intensities measured at 298 K on an Enraf-Nonius CAD-4 four-circle diffractometer (Mo $K\alpha$ radiation, graphite monochromator). Lattice parameters determined by least squares from 20 reflections ($12 \le 2\theta \le 28^\circ$). Total of 1174 reflec- $\theta = 26^{\circ}$ tions up to $(0 \le h \le 6,$ $0 \leq k \leq 10$, $-14 \le l \le 14$) measured in the ω -2 θ scan mode, 920 reflections considered as observed $[F_o > 1.5\sigma(F_o)]$. Three reference reflections monitored every hour showed no significant variation in intensity. No absorption or secondary-extinction correction. Structure solved by MULTAN11/82 (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1982) and refined by full-matrix least squares with the SDP system (B. A. Frenz & Associates Inc., 1985). Weights of each reflection in refinement (on F) calculated from $w = 1/\sigma^2(F_o)$, $\sigma(F_o)$ being the e.s.d., based on counting statistics, of the observed structure factor. Scattering factors taken from International Tables for X-ray Crystallography (1974). The total number of parameters refined was 153: one scale

factor, position parameters and anisotropic thermal parameters for non-H atoms; no attempt was made to refine the positions [calculated at d(C-H) =0.95 Å] or isotropic thermal parameters (B = 5.0 Å²) of the H atoms. Refinement resulted in final values of R = 0.046, wR = 0.049 and S = 3.18; in the last cycle $(\Delta/\sigma)_{\text{max}} = 0.02$. Final max. and min. $\Delta\rho$ were 0.21 and -0.28 e Å⁻³, respectively. All calculations performed on a MicroPDP11/73 computer. The final

Table 1. Fractional atomic coordinates (\times 10⁴) and equivalent isotropic thermal parameters $(Å^2)$ for non-H atoms

E.s.d.'s are in parentheses.

 $B_{eq} = \frac{4}{3} \left[a^2 B(1,1) + b^2 B(2,2) + c^2 B(3,3) + ab(\cos \gamma) B(1,2) + b^2 B(2,2) + c^2 B(3,3) + b^2 B(2,2) + b^2 B(2,3) + b^2 B$ $ac(\cos\beta)B(1,3) + bc(\cos\alpha)B(2,3)].$

	x	y	Z	Bea
C(1)	6743 (7)	9188 (6)	3037 (3)	2.85 (8)
O(2)	8225 (5)	7740*	2927 (2)	3.35 (6)
C(3)	8373 (8)	7438 (6)	1829 (4)	3.57 (9)
O(3)	9739 (7)	6432 (5)	1555 (3)	6.46 (9)
C(4)	6521 (8)	8443 (6)	1072 (3)	3.62 (9)
C(5)	7581 (8)	9748 (7)	362 (3)	4.1 (1)
N(6)	7212 (6)	11351 (5)	881 (3)	3.56 (7)
C(7)	9187 (8)	11937 (6)	1696 (3)	3.70 (9)
O(8)	10055 (4)	10827 (4)	2582 (2)	2.71 (5)
C(9)	8402 (7)	10615 (5)	3399 (3)	2.48 (7)
C(10)	5205 (7)	9435 (6)	1868 (3)	3.34 (9)
O(11)	5115 (5)	11094 (4)	1457 (2)	3.73 (6)
C(12)	10061 (7)	10343 (6)	4516 (3)	2.89 (8)
O(12)	8813 (5)	9580 (4)	5371 (2)	2.83 (5)
C(13)	7069 (7)	10491 (5)	5764 (3)	2.83 (8)
O(13)	6515 (5)	11839 (4)	5423 (2)	3.66 (6)
C(14)	5975 (8)	9573 (7)	6648 (3)	3.74 (9)

* Origin-defining coordinate.

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E.s.d.'s are in parentheses.

		-	
C(1)—O(2)	1.451 (5)	N(6)—O(11)	1.450 (4)
C(1)-C(9)	1.506 (6)	C(7)—O(8)	1.421 (5)
$C(1) \rightarrow C(10)$	1.543 (5)	O(8)-C(9)	1.443 (5)
O(2) - C(3)	1.348 (6)	$C(9) \rightarrow C(12)$	1.527 (5)
$C(3) \rightarrow O(3)$	1.193 (6)	$\dot{c}(0)$	1.432 (7)
C(3) - C(4)	1.509 (6)	C(12) - O(12)	1.452 (5)
C(4) - C(5)	1.528 (8)	O(12) - O(12)	1.355 (5)
C(4) - C(10)	1.512 (6)	C(13) = O(13)	1.194(5)
C(5) = N(6)	1.469 (7)	C(13) - C(14)	1.401 (6)
N(6) - C(7)	1.401(7)	C(13) $C(14)$	1471(0)
$\Pi(0) = \mathbb{C}(7)$	1 +1 (3)		
O(2)-C(1)-C(9)	108.4 (3)	N(6) C(7) O(8) 116-0 (4)
O(2)C(1)-C(10)) 105.8 (3)	C(7)-O(8)-C(9)) 113.4 (3)
C(9)-C(1)-C(10) 113.4 (4)	C(1)-C(9)-O(8)	108.6 (4)
C(1)-O(2)-C(3)	110.5 (3)	C(1) - C(9) - C(1)	2) 114.2 (3)
O(2)-C(3)-O(3)	121.5 (4)	O(8)-C(9)-C(1)	2) 104.2 (3)
O(2) - C(3) - C(4)	110.7 (4)	C(1) - C(10) - C(40)	4) 104.2 (3)
O(3) - C(3) - C(4)	127.7 (5)	C(1) - C(10) - O(11) 114·9 (4)
C(3) - C(4) - C(5)	115.1 (4)	C(4) - C(10) - O(10)	11) 106·5 (3)
C(3) - C(4) - C(10)	$105 \cdot 1$ (3)	N(6)-O(11)-C(10) 107.9 (4)
C(5) - C(4) - C(10)	103.8(5)	C(9) - C(12) - O(12)	12) 112.7(3)
C(4) - C(5) - N(6)	107.0 (4)	$C(12) \rightarrow O(12) \rightarrow C$	(13) 116.6 (3)
C(5) - N(6) - C(7)	116-1 (3)	O(12) - C(13) - O(12) - O(13) - O(13	(13) 123.0 (4)
C(5) - N(6) - O(1)	1) 1040(3)	O(12) - C(13) - C(13	(14) 111.0 (4)
C(7) - N(6) - O(1)	1) 108.3(3)	0(13) - C(13) - C	(14) 126.1 (4)
	.) 1005(5)	0(15) 0(15) 0	() .=• . (.)
C(10)-C(1)-O(2)-	-C(3) 19·6 (4)	C(5)-N(6)-O(11)-	-C(10) 34·2 (4)
C(10)-C(1)-C(9)-	-O(8) - 60.3(4)	C(7)-N(6)-O(11)-	-C(10) - 89·8 (4)
O(2)-C(1)-C(10)-	-C(4) - 18.2(4)	N(6)-C(7)-O(8)-	C(9) = 72.1(4)
C(9)-C(1)-C(10)-	-O(11) - 15·4 (5)	C(7)-O(8)-C(9)-	C(1) 91·2 (4)
C(1)O(2)C(3)	C(4) - 12·9 (5)	C(1)-C(9)-C(12)-	-O(12) - 42·7 (4)
O(2)-C(3)-C(4)-	C(10) 0·5 (5)	C(1)-C(10)-O(11)	—N(6) 86·0 (4)
C(10)-C(4)-C(5)-	-N(6) 9·1 (4)	C(4) - C(10) - O(11)	-N(6) - 28.7(4)
C(3) - C(4) - C(10) - C(10)	-C(1) 10-9 (5)	C(9) - C(12) - O(12)	-C(13) = 67.0 (4)
C(3) - C(4) - C(10) - C(10)	-U(11) 11.4 (4)	C(12) = O(12) = C(13)	D = U(13) = 1.7(5)
U(4) - U(3) - N(6) - O(11)	O(11) = 20.0(4) O(2) = 64.2(4)	U(12) - U(12) - U(13)	-1/8.9(3)
U(II)-N(0)-U(/)-			

atomic coordinates are given in Table 1.* Bond distances, bond angles and torsion angles are given in Table 2. Atom numbering is shown in Fig. 1, unit-cell contents in Fig. 2.

Related literature. The organic synthesis and spectroscopic data for the title compound have been reported (Panfil, Bełżecki, Chmielewski & Suwińska, 1989).



Fig. 1. ORTEPII (Johnson, 1976) view of the molecules with the atom numbering. The thermal ellipsoids are given at 50% probability.



Fig. 2. Packing diagram.

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^{*} Lists of structure factors, anisotropic thermal parameters for the non-H atoms, positional and isotropic thermal parameters for the H atoms and a table of the shortest non-bonded contacts have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 51964 (14 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.